

Global Reaction Route Mapping on the Potential Energy Surface



via the Anharmonic Downward Distortion Following

Koichi Ohno Graduate School of Science, Tohoku University, JAPAN

Global Mapping of the World



Ptolemaios's map



Discovery of New Routes

1492 Christpher Columbus1498 Vasco da Gama1497 Sebastien Gabbot1499 Amerigo Vespucci1519 Ferdinand Magellan

Captain James Cook 1728-1779



Exploration of the Unknown World!



How to Explore the Chemical World?

Global Map? Tool for Exploration?

Fundamental Problems of Chemistry?

for a given Chemical Formula, such as C_NH_MO_L. **1) What kinds of chemical species (isomers) exist? 2) How are they converted each other?**

3) How do they dissociate into smaller species?

o solve These Problems Theoretically,

Global Mapping of Potential Energy Surfaces (PES s required to be made.

Vinima:

Equilibrium Structures (EQ)

Saddle points:

Fransition Structures (7S)

valleys:

ntrinsic Reaction Coordinates (IRC)





yields Potential Energy Surface.

Schrödinger / Heisenberg / Dirac · Pople / Kohn · Nakatsuji

In order to study PES, store 24 you need to solve Schrödinger's equation, many ··· times!

Algorithms for PES Mapping

EQ: Geometry Optimization Problem Depend on the initial guess! Numerous Try-&-Errors! EQ TS IRC IRC EQ EQ

EC

FO

TS: Transition Structure Search Eigen Vector Following (*EVF*) : Cerjan & Miller (1981) SEAM : Jensen (1992)

BB : Floudas et al. (1992)

Gradient Extremal (*GE*) : Sun & Ruedenberg (1993) Sphere Optimization (*SO*) : Abashkin & Russo (1994) Nudged Elastic Band (*NEB*) : Jonsson et al. (1998) *Problem* Partially possible, but *generally impossible!*

IRC: Reaction Pathway Search Downhill walks from TS are easy along steepest decent paths! Problem Uphill walks along reaction paths are impossible!

No Algorithm for Uphill Walk along Reaction Path!

Uphill walks along reaction paths make it possible to perform Global Reaction Route Mapping without wondering on PES!



How to find reaction routes from EQ?



Anharmonic Downward Distortion (ADD) indicates Reaction Routes!

ADD : Compass to the Unknown Chemical World)

How to Find ADDs?

- Use the <u>Iso-Energy Harmonic Hyper-Surface</u>.
- Scaled Normal Coordinate : $q_i = \frac{1}{2}Q_i$



Optimization on Sphere Surface!

How to Find & Follow ADD?

- ADD can be found as real energy minima on a scaled hyper-sphere.
- Reaction pathways can be followed by expanding the size of hyper-spheres.



Scaled Hypersphere Search(SHS) Method

K.Ohno & S.Maeda, Chem. Phys. Lett. 384, 277 (2004).

GRRM1.00 is the first computer program based on the **SHS** (**ADD** following) algorithm for an automated exploration of chemical reaction pathways by using energies obtained from solutions of H = E

- **GRRM1.00** copes with long standing fundamental problems in chemistry by automated exploration of chemical reaction pathways.
 - GRRM1.00 automatically explores unknown isomers.
- **GRRM1.00** automatically explores unknown synthetic routes.
- GRRM1.00 automatically explores unknown dissociation channels.
- *GRRM***1.00** develops an unexplored world of chemistry by elucidating unknown chemical reaction networks.
- GRRM1.00 is useful for production of the Atlas for the chemical world.
- **GRRM1.00** is useful for design of new chemical compounds and reactions.
- *GRRM*1.00 is useful for designing new tactics for energy/environment problems.
- GRRM1.00 is useful for elucidation of catalysis and design of new catalysts.
- **GRRM1.00** is an epoch-making program of potential analyses for the following problems.
- **Normal coordinate analysis** Normal coordinate calculations can be made at arbitrary structures. Optionally, enthalpy and Gibbs energies can also be obtained.
- **Optimization of equilibrium structures** Equilibrium structures can be optimized by **SIRFO** and **BFGS** methods.
- **Optimization of transition structures** Transition structures can be optimized by **SIRFO** and **Bofill's** methods.
- **IRC search IRC** can be traced by **Page** and **McIver** methods.
- **GRRM search** Global reaction route mapping (**GRRM**) can be made for the potential surface of a given chemical formula. Starting from an equilibrium structure, automated search of dissociation and isomerization can be performed to explore **GRRM** corresponding to the Atlas of chemical reaction routes. Optionally, exploration of reaction routes can be made for the limited region around a particular structure.
- **One step TS search** An efficient search of the reaction pathway connecting a reactant and a product can be made to determine the transition structure (**TS**). This procedure can be done automatically without initial guess, and this technique is much more rapid and applicable than any other methods, such as the **NEB** method.
- Intermediate search Intermediates between a pair of isomers can be found, even if they are far apart. The SHS method in the hypersphere-contraction-mode enables us to explore multi-step reaction pathways, even if they amount to several tens of steps.
- **Program Package & Requirement for GRRM1.00**
- GRRM1.00 utilizes energies obtained by Gaussian03.
- **GRRM1.00**. a 64-bit or a 32-bit version. can be used under a Linux/Unix environment.
- *GRRM***1.00** can be used for research and education, after application to the following address by E-mail. ohnok@gpcrkk.chem.tohoku.ac.jp
- Further information for *GRRM***1.00** is available at the following page.
- http://gpcrkk.chem.tohoku.ac.jp/

GRRM for H₂CO

5 EQ 9 TS



Gradient Extremal Method(HF/STO-3G): Bodensgard & Jensen, J. Chem. Phys. 104, 8025 (1996).

Reduced Gradient Following Method(HF/STO-3G): Quapp et al. J.Compt.Chem. 19, 1087 (1998).

Scaled Hypersphere Search Method(MP2/3-21G): Ohno & Maeda, Chem. Phys. Lett. 384,277(2004).

Scaled Hypersphere Search Method(B3LYP/6-31G): Maeda & Ohno, J.Phys.Chem.A 109,5742(2005).

GRRM for H₂CO₂ 13 EQ 30 TS **Peroxide** Acid TS0/D (C1) EQ0 (Cs) TS0/10 (C1) EQ10 (C1) TS6/10 (C1) EQ6 (Cs) TS6/D (Cs) 291.2 kJ mol⁻¹ 0.0 kJ mol 708.8 kJ mol 713.3 kJ mol⁻¹ 718.0 kJ mol⁻¹ 463.3 kJ mol⁻¹ 459.7 k.Lmol H₂O + CO H2O + CO 53.0 kJ mol 53.0 kJ mol⁻¹ TS6/12 (C1) H2 + CO2 881.0 kJ mol⁻¹ 3.4 kJ mol⁻¹ TS3/D (C1) TS0/3 (Cs) TS8/10 (C1) 724.1 kJ mol⁻¹ TS5/D (C2v) 337.5 kJ mol⁻¹ 333.1 kJ mol TS6/8 (C1) 567.0 kJ mo TS0/1 (C1) 742.9 kJ mol 55.3 kJ mol⁻¹ TS6/11 (Cs) EQ3 (Cs) 842.2 kJ mol TS5/10 (C1) 181.7 kJ mol TS1/8 (Cs) EQ1 (Cs) 743.3 kJ mol TS8/12 (Cs 19.1 kJ mol 512.6 kJ mol⁻¹ 871.2 kJ mt H2 + CO2 EQ5 (C2v) 3.4 kJ mol EQ12 (Cs) 397.6 kJ mol TS10/11 (C1) 820.9 k.l mol TS1/D (Cs) 835.4 kJ mo EQ8 (Cs) 307.3 kJ mol 510.9 kJ mol TS4/D (C2v) TS7/12 (C1) S11/12 (Cs) TS2/3 (C1) 346.1 kJ mol⁻¹ 950.8 kJ mol⁻¹ 995.6 kJ mol⁻¹ 251.1 kJ mol⁻¹ TS1/2 (Cs) TSB/9 (C1) TS3/4 (C1) 335.4 kJ mol¹ 523.9 kJ mol TS7/11 (C1) 259.3 kJ mol⁻¹ TS5/7 (C1) 887.7 kJ mol" 577.3 kJ mol1 O11 (Cs) 779.4 kJ mol TS2/11 (Cs) 913.3 kJ mol TS2/4 (Cs) EQ2 (C2v) EQ9 (Cs) TS7/9 (Cs) EQ7 (Cs) EQ4 (C2v) 486.5 k.l mol 216.0 kJ mol 653.4 kJ mol 177.5 kJ mol 523.7 kJ mol 636.4 kJ mol

K. Ohno, S. Maeda, J. Phys. Chem. A, 110, 8933 (2006).

GRRM for C₂H₃N

17 EQ 59 TS 9 EQ & 41 TS were newly found!



X. Yang, S. Maeda, and K. Ohno, *J. Phys. Chem. A* 109, 7319 (2005)

Application to finding a new synthetic route with no byproducts recommended by economy and environments

- Find Dissociation Channels starting from the aimed compound (A)
 - A B+C
- Trace back to the initial compounds (A) from the dissociation products (*B* and *C*)





Multi-step Conversion Routes of C20 Clusters



S Maada and K Ohna I Cham Dhua 124 174206 (2006

D-L Conversion Routes for Optical Isomers







K. Ohno and S. Maeda, *Chem. Letters* 35, 492 (2006)

Large ADD Following

Full ADD (FADD) Full CI
Large ADD (LADD) Limited CI

 Large ADD finds Lower Barrier leading to Lower EQ.

Lower Energy Regions connected by Lower Barriers can be Searched Efficiently.



Thermodynamic Analyses for Clusters : $(H_2O)_8$



S. Maeda and K. Ohno, J. Phys. Chem. A 111, 4527 (2007)

$H^{+}(H_{2}O)_{8}$



Micro Solvation of H₂S in Water



7-EQ1(C,)

7-EQ6(C₁)

 $\Delta E = 7.3 \text{ kJ/mol}$

 $\Delta G = 0.9 \text{ kUmol}$

 $7-EQU(C_1)$

 $\Delta E = 13.2 \text{ kJ/mol}$

 $\Delta G = 0.0 \text{ kJ/mol}$

 $\Delta E = 13.4 \text{ kJ/mol}$

 $\Delta G = 1.7 \text{ kJ/mol}$

 $\Delta E = 0.0 \text{ kJ/mol}$

 $\Delta G = 1.0 \text{ kJ/mol}$











7-EQ2(C1) $\Delta E = 4.3 \text{ kJ/mol}$ $\Delta G = 3.6 \text{ kJ/mol}$

7-EO7(C1)

 $\Delta E = 8.7 \text{ kJ/mol}$

 $\Delta G = 0.8 \text{ kJ/mol}$

7-EQ3(C1) $\Delta E = 5.5 \text{ kJ/mol}$ $\Delta G = 3.7 \text{ kJ/mol}$

7-EQ8(C₁)

7-EQ6(C₁)

 $\Delta E = 7.3 \text{ kJ/mol}$

 $\Delta G = 0.9 \text{ kJ/mol}$

 $\Delta E = 9.7 \text{ kJ/mol}$

 $\Delta G = 1.7 \text{ kJ/mol}$

J. Phys. Chem. A 111,13168 (2007)

 $\Delta E = 9.7 \text{ kJ/mol}$

 $\Delta G = 1.7 \text{ kJ/mol}$

7-EQ4(C,) $\Delta E = 5.8 \text{ kJ/mol}$ $\Delta G = 4.3 \text{ kJ/mol}$

7-EO5(C₁) $\Delta E = 7.1 \text{ kJ/mol}$ $\Delta G = 4.7 \text{ kJ/mol}$

7-E09(C₁) $\Delta E = 10.1 \text{ kJ/mol}$ $\Delta G = 1.5 \text{ kUmol}$

7-EO10(C₁) AE = 10.3 kJ/mol $\Delta G = 2.6 \text{ kJ/mol}$



7-EQ7(C1)

 $\Delta E = 8.7 \text{ kJ/mol}$

 $\Delta G = 0.8 \text{ kJ/mol}$

 $\Delta E = 25.0 \text{ kJ/mol}$

 $\Delta G = 1.7 \text{ kJ/mol}$

S. Maeda and K. Ohno,





7-EQ1(C,) $\Delta E = 0.0 \text{ kJ/mol}$ $\Delta G = 1.0 \text{ kJ/mol}$

7-EQ9(C1) $\Delta E = 10.1 \text{ kJ/mol}$ $\Delta G = 1.5 \text{ kJ/mol}$



 $7-EO8(C_1)$



7-EQ14(C₁) $\Delta E = 28.2 \text{ kJ/mol}$ $\Delta G = 1.7 \text{ kJ/mol}$

7-EQ15(C1) $\Delta E = 10.4 \text{ kJ/mol}$ $\Delta G = 1.7 \text{ kJ/mol}$

Asymmetric Catalyst : RuHCI-BINAP

 $CH_{3}COCH_{2}COOCH_{3} + H_{2}$ $CH_{3}C^{*}HOHCH_{2}COOCH_{3}$



(A)

TS1(R)

0.0 kJ/mol

TS2 (S) 15.4 kJ/mo

A New GRRM for H₂CO



(0.0 kJ/mol)

CO + H₂

16.26 kJ/mo

(20.54 kJ/mo

alanaa 206 1150 (2001)

Summary

Anharmonic Downward Distortion (ADD) has been discovered as a Compass indicating Directions of Reaction Paths.

Uphill-Walks along Reaction Paths become possible by following ADD.

Scaled Hypersphere Search (SHS) Algorithm has been developed for Global Reaction Route Mapping on the Potential Energy Surface (PES).

Uphill-Walks on PES enables us to discover Unexplored Chemistry.



Collaborators

O for this work



A View of a Saddle Point near Mt **ZAO** in JAPAN